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USACorrespondence e-mail:
potenza@rutchem.rutgers.edu**Key indicators**Single-crystal X-ray study
 $T = 296$ K
Mean $\sigma(\text{C}-\text{C}) = 0.004$ Å
 R factor = 0.063
 wR factor = 0.185
Data-to-parameter ratio = 10.5For details of how these key indicators were
automatically derived from the article, see
<http://journals.iucr.org/e>.**A second polymorph of 4,5-diphenyl-1*H*-imidazole**

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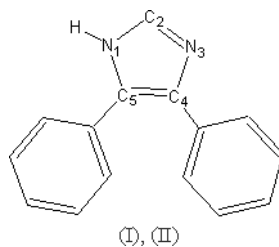
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The title compound, $\text{C}_{15}\text{H}_{12}\text{N}_2$, (II)₍₄₉₉₎, contains two molecules in the asymmetric unit and is a polymorphic form of a previously reported structure, (I)₍₄₉₅₎, of the same molecule. The molecules in both polymorphs exhibit nearly identical bond distances, but show significantly different phenyl/imidazole dihedral angles and pack differently in the crystalline state. In each polymorph, strong hydrogen-bonded polymer chains, involving the N atoms of the imidazole fragments in adjacent molecules, are formed.

Comment

Substituted 4,5-diphenylimidazoles are known to be active pharmacologically. For example, a mevalonate derivative containing a 4,5-diphenyl-1*H*-imidazole group has been shown to reduce cholesterol levels in rats (Harris *et al.*, 1992). From the standpoints of drug performance and of intellectual property, polymorphism is an important aspect of pharmacologically active compounds. As an example, sertraline hydrochloride, the active drug in Zoloft[™], has been found to have 17 different polymorphic forms (Almarsson *et al.*, 2003). One of our current interests is to use monodentate imidazoles and benzimidazoles as building blocks to synthesize bis-chelating ligands which are useful for preparing metal-ion complexes with unusual coordination geometries and properties (Stibrany *et al.*, 2004). Several methods have been reported for joining imidazole fragments to prepare such ligands (Gorun *et al.*, 1996; Sugimoto *et al.*, 1998). In this study, we report a second polymorph of the monodentate building block 4,5-diphenylimidazole.



The structure of (II)₍₄₉₉₎ contains two 4,5-diphenyl-1*H*-imidazole molecules in the asymmetric unit (Fig. 1), in contrast to the polymorph reported previously, (I)₍₄₉₅₎, which contains only one such molecule [refcode OCUSUA (Stibrany *et al.*, 2001) in the Cambridge Structural Database (Version 5.24; Allen, 2002)]. Here, we use the nomenclature encouraged by Bernstein (2002), in which polymorphs are designated in order of discovery by increasing Roman numerals followed by their melting point in Kelvin. Bond lengths in the imidazole fragments of (I)₍₄₉₅₎ and (II)₍₄₉₉₎ agree well (Table 2). In addition,

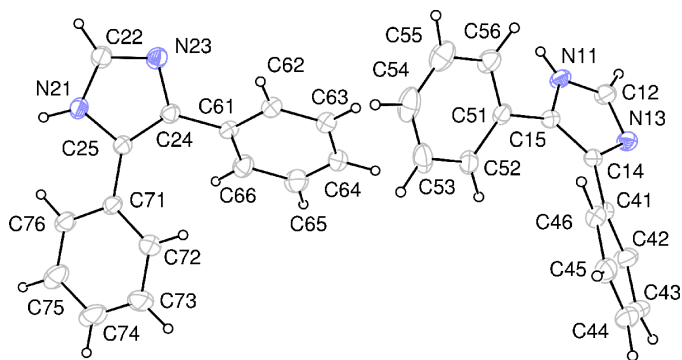


Figure 1
The contents of the asymmetric unit of (II)₍₄₉₉₎, showing 25% probability displacement ellipsoids and the atom-numbering scheme.

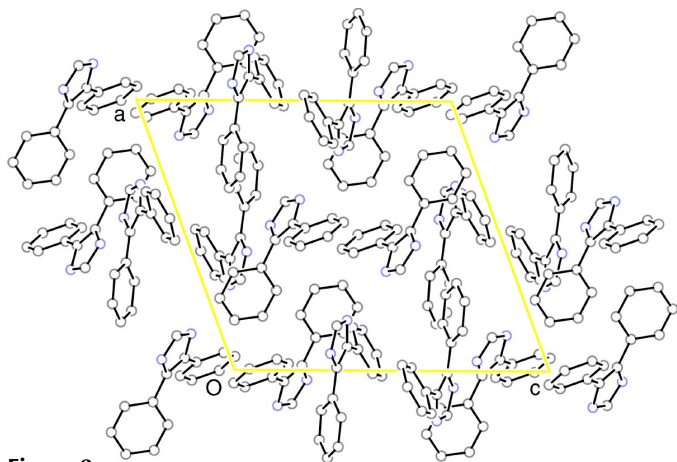


Figure 2
View, along the *b* axis, of the contents of the unit cell of (II)₍₄₉₉₎. H atoms have been omitted for clarity.

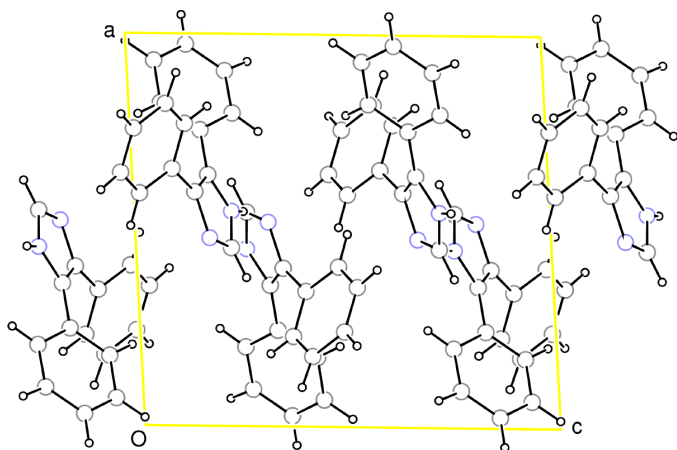


Figure 3
View, along the *b* axis, of the contents of the unit cell of (I)₍₄₉₅₎.

the imidazole (im) and phenyl (ph) fragments in both polymorphs are planar to within 0.01 Å, as expected. From a molecular perspective, the polymorphs differ primarily in the relative orientation of the ph and im fragments. In (II)₍₄₉₉₎, the ph/im dihedral angles are 35.81 (8) and 42.71 (8)° for one molecule, and 34.22 (11) and 41.15 (12)° for the second unique molecule. A greater difference in these angles is observed for polymorph (I)₍₄₉₅₎, in which the phenyl rings are tilted by 29.79 (6) and 42.75 (8)° with respect to the imidazole plane.

In crystal structure of (II)₍₄₉₉₎ (Fig. 2), one of the unique molecules (*A*) is related to the others by centers of symmetry at $(0, \frac{1}{2}, 0)$ and by the *n*-glide plane symmetry operations. The second unique molecule (*B*) is related similarly by centers of symmetry at $(\frac{1}{2}, \frac{1}{2}, 0)$ and by the *n*-glide plane operations. When coupled with translations along the *b* direction, a structure results which may be viewed as consisting of slightly overlapping, alternating slabs of *A* and *B* molecules parallel to the ($\bar{1}01$) planes. *A*- and *B*-type molecules are linked to each other along the [101] direction by strong N—H...N hydrogen bonds (Table 1) to form infinite —*A*—*B*— chains. Intermolecular C—C and C—N distances suggest that the molecules are otherwise linked primarily by C—H... π interactions, with little, if any, π stacking.

In crystals of polymorph (I)₍₄₉₅₎ (space group *P*2₁/*c*, one molecule per asymmetric unit; Fig. 3), molecules related by the screw axes form columns along the *b* direction which are related to each other by the centers of symmetry and by the *c*-glide-plane symmetry operations. Within a given column, adjacent molecules are linked by N—H...N hydrogen bonds to form infinite chains along *b*. As with polymorph (II)₍₄₉₉₎, intermolecular interactions exclusive of hydrogen bonds appear to be dominated by C—H... π interactions, with little evidence to suggest π stacking.

Experimental

4,5-Diphenyl-1*H*-imidazole was synthesized by the benzoin condensation method, as reported by Bredereck & Theillig (1953). In that report, a melting point of 504 K was obtained for the product, which was recrystallized from diethyl malonate. Polymorph (I)₍₄₉₅₎ was obtained by recrystallization from 2-propanol, while the current polymorph, (II)₍₄₉₉₎, was obtained by recrystallizing (I)₍₄₉₅₎ from acetonitrile (m.p. 499 K).

Crystal data

C₁₅H₁₂N₂
M_r = 220.27
 Monoclinic, *P*2₁/*n*
a = 13.3716 (7) Å
b = 13.0984 (8) Å
c = 14.6909 (12) Å
 β = 110.343 (4)°
V = 2412.6 (3) Å³
Z = 8
D_x = 1.213 Mg m⁻³
D_m = 1.20 (1) Mg m⁻³

D_m measured by flotation in a mixture of carbon tetrachloride and cyclohexane
 Mo *K*α radiation
 Cell parameters from 960 reflections
 θ = 3.0–23.6°
 θ_{\max} = 25.0°
 μ = 0.07 mm⁻¹
T = 296 (1) K
 Prism, colorless
 0.55 × 0.44 × 0.26 mm

Data collection

Bruker SMART CCD area-detector diffractometer
 φ and ω scans
 Absorption correction: multi-scan (*SADABS*; Blessing, 1995)
 T_{\min} = 0.783, T_{\max} = 0.980
 15100 measured reflections

4234 independent reflections
 3444 reflections with $I > 2\sigma(I)$
 R_{int} = 0.037
 θ_{\max} = 25.0°
 $h = -15 \rightarrow 15$
 $k = -15 \rightarrow 13$
 $l = -17 \rightarrow 9$

Refinement

Refinement on F^2
 $R[F^2 > 2\sigma(F^2)] = 0.063$
 $wR(F^2) = 0.185$
 $S = 1.00$
 4234 reflections
 403 parameters
 All H-atom parameters refined

$w = 1/[\sigma^2(F_o^2) + (0.1214P)^2 + 0.4106P]$
 where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\max} < 0.001$
 $\Delta\rho_{\max} = 0.19 \text{ e } \text{Å}^{-3}$
 $\Delta\rho_{\min} = -0.16 \text{ e } \text{Å}^{-3}$

Table 1

Hydrogen-bonding geometry (Å, °).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
N21—H21N \cdots N13 ⁱ	0.86 (3)	2.09 (3)	2.917 (3)	160 (3)
N11—H11N \cdots N23 ⁱⁱ	0.83 (3)	2.05 (3)	2.884 (3)	178 (3)

Symmetry codes: (i) $1+x, y, 1+z$; (ii) $x-\frac{1}{2}, \frac{3}{2}-y, z-\frac{1}{2}$.

Table 2

Bond lengths (Å) in the imidazole fragments of polymorphs (I) and (II) (molecules *A* and *B*).

	(I)	(II), molecule <i>A</i>	(II), molecule <i>B</i>
N1—C2	1.340 (2)	1.334 (3)	1.336 (3)
C2—N3	1.314 (2)	1.317 (3)	1.312 (3)
N3—C4	1.385 (2)	1.391 (3)	1.383 (3)
C4—C5	1.374 (2)	1.373 (3)	1.369 (3)
N1—C5	1.376 (2)	1.378 (3)	1.378 (3)

Data collection: *SMART-WNT*(2000) (Bruker, 2000); cell refinement: *SAINT-Plus* (Bruker, 2000); data reduction: *SAINT-Plus*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1990); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *ORTEPIII* (Burnett & Johnson, 1996) and *ORTEP-32* (Farrugia, 1997); software used to prepare material for publication: *SHELXTL* (Bruker, 2000).

References

- Allen, F. H. (2002). *Acta Cryst.* **B58**, 380–388.
- Almarsson, O., Hickey, M. B., Peterson, M. L., Morissette, S. L., Soukasene, S., McNulty, C., Tawa, M., MacPhee, J. M. & Remenar, J. F. (2003). *Cryst. Growth Des.* **3**, 927–933.
- Bernstein, J. (2002). *Polymorphism in Molecular Crystals, IUCr Monographs on Crystallography*, No. 14, pp. 8–9. Oxford University Press.
- Blessing, R. H. (1995). *Acta Cryst.* **A51**, 33–38.
- Bredereck, H. & Theillig, G. (1953). *Chem. Ber.* **86**, 88–96.
- Bruker (2000). *SHELXTL* (Version 6.10), *SAINT-Plus* (Version 6.02) and *SMART-WNT/2000* (Version 5.622). Bruker AXS Inc., Madison, Wisconsin, USA.
- Burnett, M. N. & Johnson, C. K. (1996). *ORTEPIII*. Report ORNL-6895. Oak Ridge National Laboratory, Tennessee, USA.
- Farrugia, L. J. (1997). *J. Appl. Cryst.* **30**, 565.
- Goron, S. M., Stibrany, R. T., Katritzky, A. R., Slawinski, J. J., Faid-Allah, H. & Brunner, F. (1996). *Inorg. Chem.* **35**, 3–4.
- Harris, N. V., Smith, C., Ashton, M. J., Bridge, A. W., Bush, R. C., Coffee, E. C. J., Dron, D. I., Harper, M. F., Lythgoe, D. J., Newton, C. G. & Riddell, D. (1992). *J. Med. Chem.* **35**, 4384–4392.
- Sheldrick, G. M. (1990). *Acta Cryst.* **A46**, 467–473.
- Sheldrick, G. M. (1997). *SHELXL97*. University of Göttingen, Germany.
- Stibrany, R. T., Potenza, J. A. & Schugar, H. J. (2001). Private Communication (reference number CCDC 172750) to the Cambridge Structural Database, 12 Union Road, Cambridge, England.
- Stibrany, R. T., Lobanov, M. V., Schugar, H. J. & Potenza, J. A. (2004). *Inorg. Chem.* **43**, 1472–1480.
- Sugimoto, H., Nagayama, T., Maruyama, S., Fujinami, S., Yasuda, Y., Suzuki, M. & Uehara, A. (1998). *Bull. Chem. Soc. Jpn.* **71**, 2267–2279.